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6-Methyl-2-phenyl-4,5-dihydropyridazin-3(2H)-one

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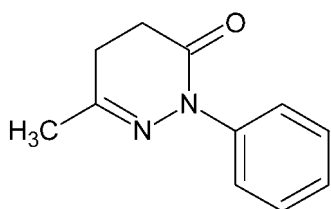
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.032; wR factor = 0.094; data-to-parameter ratio = 9.5.

In the title molecule, $\text{C}_{11}\text{H}_{12}\text{N}_2\text{O}$, the pyridazine ring has a skew-boat conformation. The dihedral angle between the phenyl ring [r.m.s deviation = 0.0039 (15) Å] and the best mean-plane of the pyridazine ring [r.m.s deviations = 0.2629 (15) Å] is 53.27 (10)°. In the crystal, molecules are linked by $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds and $\text{C}-\text{H}\cdots\pi$ interactions involving the methyl group and the phenyl ring of a symmetry-related molecule.

Related literature

For the similar structure, 2-(4-methoxyphenyl)-6-(trifluoromethyl)-4,5-dihydropyridazin-3(2H)-one, see: Wan *et al.* (2009). For conformation analysis of six-membered rings, see: Cremer & Pople (1975).



Experimental

Crystal data

 $\text{C}_{11}\text{H}_{12}\text{N}_2\text{O}$ $M_r = 188.23$

Monoclinic, $P2_1$
 $a = 6.4151$ (2) Å
 $b = 7.9010$ (2) Å
 $c = 10.1888$ (3) Å
 $\beta = 106.607$ (1)°
 $V = 494.89$ (2) Å³

$Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 296$ K
 $0.24 \times 0.15 \times 0.12$ mm

Data collection

Bruker APEXII CCD detector
 diffractometer
 7678 measured reflections

1220 independent reflections
 1154 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.094$
 $S = 1.06$
 1220 reflections
 129 parameters

1 restraint
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.15$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.13$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C1–C6 ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C3}-\text{H3}\cdots\text{O1}^{\text{i}}$	0.93	2.54	3.371 (3)	149
$\text{C5}-\text{H5}\cdots\text{O1}^{\text{ii}}$	0.93	2.50	3.346 (2)	152
$\text{C8}-\text{H8B}\cdots\text{O1}^{\text{iii}}$	0.97	2.55	3.474 (3)	159
$\text{C11}-\text{H11B}\cdots\text{Cg1}^{\text{iv}}$	0.96	2.89	3.812 (3)	161

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z + 2$; (ii) $x - 1, y, z$; (iii) $-x + 1, y - \frac{1}{2}, -z + 1$; (iv) $-x, y + \frac{1}{2}, -z + 1$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

The authors thank the Unit of Support for Technical and Scientific Research (UATRS, CNRST) for the X-ray measurements.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2275).

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supplementary materials

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6-Methyl-2-phenyl-4,5-dihydropyridazin-3(2H)-one

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Comment

The molecular structure of the title molecule is illustrated in Fig. 1. The bond distances and angles are similar to those reported for 2-(4-Methoxyphenyl)-6-(trifluoromethyl)-4,5-dihydropyridazin-3(2H)-one (Wan *et al.*, 2009). The pyridazine ring has a skew-boat conformation; Puckering Amplitude (Q) = 0.428 (2) Å, $\theta = 69.9$ (2)°, $\phi = 207.6$ (3)° (Cremer and Pople, 1975). The dihedral angle between the phenyl ring and the best mean-plane of the pyridazine ring (r.m.s deviations: 0.2629 (15) Å) is 53.27 (10)°.

In the crystal molecules are linked *via* non-classical C—H...O hydrogen bonds (Table 1), forming a two-dimensional network (Fig. 2). Molecules are also linked by C—H... π interactions involving a methyl H-atom and the phenyl ring of a symmetry related molecule (Table 1).

Experimental

A mixture of phenylhydrazine (2.7 ml, 27 mmol) and levulinic acid (2.6 ml, 25 mmol) in 60 ml of ethanol were refluxed for 4 h. After cooling the reaction mixture was poured onto ice. The solid obtained was filtered off and recrystallized from methanol to give the title compound as colourless crystals: Yield 3.9 g (85%); Mp: 367–369 K. Spectroscopic data for the title compound is given in the archived CIF.

Refinement

In the final cycles of refinement, in the absence of significant anomalous scattering effects, 906 Friedel pairs were merged and Δf set to zero. H-atoms were positioned geometrically, with C—H = 0.93 Å for CH(aromatic), 0.97 Å for CH₂ and 0.96 Å for CH₃ H-atoms. They were constrained to ride on their parent atoms, with $U_{iso}(H) = k \times U_{eq}(C)$, where $k = 1.5$ for CH₃ H-atoms and $k = 1.2$ for all other H-atoms.

Figures

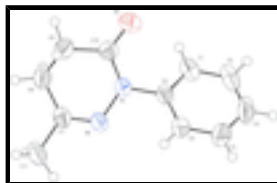


Fig. 1. A view of the molecular structure of the title molecule, showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.

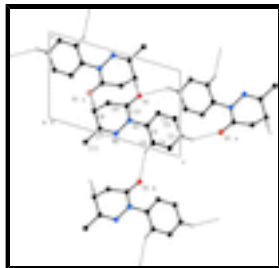


Fig. 2. A view along the *b* axis of the crystal packing of the title compound, showing the chain formed by C—H...O interactions (dashed lines; see Table 1 for details). H-atoms not involved in hydrogen bonding have been omitted for clarity.

6-Methyl-2-phenyl-4,5-dihydropyridazin-3(2*H*)-one

Crystal data

$C_{11}H_{12}N_2O$

$M_r = 188.23$

Monoclinic, $P2_1$

Hall symbol: P 2yb

$a = 6.4151(2) \text{ \AA}$

$b = 7.9010(2) \text{ \AA}$

$c = 10.1888(3) \text{ \AA}$

$\beta = 106.607(1)^\circ$

$V = 494.89(2) \text{ \AA}^3$

$Z = 2$

$F(000) = 200$

$D_x = 1.263 \text{ Mg m}^{-3}$

Melting point: 397 K

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 256 reflections

$\theta = 2.4\text{--}26.5^\circ$

$\mu = 0.08 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Prism, colourless

$0.24 \times 0.15 \times 0.12 \text{ mm}$

Data collection

Bruker APEXII CCD detector
diffractometer

Radiation source: fine-focus sealed tube

graphite

ω and φ scans

7678 measured reflections

1220 independent reflections

1154 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.031$

$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 2.1^\circ$

$h = -8 \rightarrow 7$

$k = -8 \rightarrow 10$

$l = -12 \rightarrow 13$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.032$

$wR(F^2) = 0.094$

$S = 1.06$

1220 reflections

129 parameters

1 restraint

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0582P)^2 + 0.0443P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.13 \text{ e \AA}^{-3}$

Extinction correction: *SHELXL*,

$F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Primary atom site location: structure-invariant direct methods Extinction coefficient: 0.21 (2)

Special details

Experimental. Spectroscopic data for the title compound: $^1\text{H-NMR}$ (CDCl_3): δ 2.14 (s, 3H, CH₃), 2.54–2.65 (m, 4H, –CH₂–CH₂–), 7.20–7.31 (m, 2H, H–Ar), 7.36–7.40 (m, 1H, H–Ar), 7.48–7.51 (m, 2H, H–Ar); $^{13}\text{C-NMR}$ (CDCl_3): δ 23.2 (CH₃), 26.3 (CH₂), 27.7 (CH₂), 125 (2CH–Ar), 126.5 (CH–Ar), 128.6 (2CH–Ar), 141.1, 154 (2 C), 165 (C=O).

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.4300 (3)	0.0477 (3)	0.83092 (17)	0.0488 (5)
C10	0.4122 (2)	0.2873 (2)	0.59455 (17)	0.0413 (4)
C11	–0.1543 (3)	0.1310 (3)	0.2779 (2)	0.0598 (6)
C2	0.4258 (4)	–0.0098 (3)	0.95947 (18)	0.0608 (6)
C3	0.2448 (4)	0.0173 (4)	1.00372 (19)	0.0673 (7)
C4	0.0679 (4)	0.1005 (4)	0.92101 (19)	0.0652 (6)
C5	0.0682 (3)	0.1580 (3)	0.79218 (17)	0.0488 (4)
C6	0.2502 (3)	0.1303 (2)	0.74779 (15)	0.0389 (4)
C7	0.0504 (3)	0.1740 (2)	0.38723 (16)	0.0416 (4)
C8	0.2420 (3)	0.2321 (3)	0.34560 (17)	0.0513 (4)
C9	0.3773 (3)	0.3520 (3)	0.45107 (18)	0.0527 (5)
H1	0.5527	0.0307	0.8012	0.059*
H11A	–0.1265	0.0394	0.2233	0.090*
H11B	–0.2028	0.2282	0.2207	0.090*
H11C	–0.2646	0.0979	0.3195	0.090*
H2	0.5454	–0.0666	1.0154	0.073*
H3	0.2425	–0.0207	1.0897	0.081*
H4	–0.0537	0.1187	0.9516	0.078*
H5	–0.0522	0.2143	0.7365	0.059*
H8A	0.1937	0.2887	0.2576	0.062*
H8B	0.3295	0.1353	0.3359	0.062*
H9A	0.5173	0.3679	0.4341	0.063*
H9B	0.3053	0.4611	0.4420	0.063*
N1	0.25017 (19)	0.18561 (19)	0.61361 (12)	0.0389 (3)
N2	0.0518 (2)	0.1529 (2)	0.51137 (14)	0.0434 (4)
O1	0.5696 (2)	0.3270 (2)	0.68880 (14)	0.0560 (4)

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0539 (9)	0.0478 (11)	0.0389 (8)	0.0057 (8)	0.0038 (7)	-0.0021 (8)
C10	0.0413 (7)	0.0409 (9)	0.0420 (8)	-0.0027 (7)	0.0125 (6)	-0.0036 (7)
C11	0.0667 (12)	0.0614 (14)	0.0393 (9)	-0.0110 (11)	-0.0044 (8)	0.0058 (9)
C2	0.0803 (13)	0.0507 (11)	0.0362 (8)	0.0005 (11)	-0.0081 (8)	0.0020 (9)
C3	0.0981 (17)	0.0708 (15)	0.0300 (7)	-0.0190 (14)	0.0136 (9)	0.0020 (9)
C4	0.0708 (12)	0.0870 (17)	0.0420 (9)	-0.0164 (12)	0.0230 (9)	-0.0045 (11)
C5	0.0465 (8)	0.0620 (11)	0.0374 (8)	-0.0036 (9)	0.0112 (6)	0.0008 (8)
C6	0.0448 (8)	0.0387 (8)	0.0301 (7)	-0.0042 (7)	0.0056 (6)	-0.0015 (6)
C7	0.0499 (8)	0.0360 (8)	0.0346 (7)	-0.0007 (7)	0.0048 (6)	0.0022 (7)
C8	0.0625 (10)	0.0570 (11)	0.0352 (7)	-0.0039 (9)	0.0151 (7)	0.0019 (8)
C9	0.0579 (10)	0.0550 (11)	0.0474 (9)	-0.0115 (9)	0.0189 (8)	0.0028 (9)
N1	0.0390 (6)	0.0451 (8)	0.0306 (6)	-0.0027 (6)	0.0067 (5)	0.0004 (6)
N2	0.0416 (7)	0.0494 (9)	0.0342 (6)	-0.0062 (7)	0.0027 (5)	0.0034 (6)
O1	0.0454 (6)	0.0631 (9)	0.0533 (7)	-0.0136 (7)	0.0044 (5)	-0.0029 (7)

Geometric parameters (\AA , $^\circ$)

C1—H1	0.9300	C5—C4	1.389 (3)
C1—C2	1.394 (3)	C6—N1	1.4351 (19)
C10—C9	1.504 (2)	C6—C5	1.385 (2)
C10—N1	1.371 (2)	C6—C1	1.384 (2)
C10—O1	1.219 (2)	C7—C11	1.498 (2)
C11—H11C	0.9600	C7—C8	1.483 (2)
C11—H11B	0.9600	C7—N2	1.273 (2)
C11—H11A	0.9600	C8—H8B	0.9700
C2—H2	0.9300	C8—H8A	0.9700
C2—C3	1.377 (4)	C9—H9B	0.9700
C3—H3	0.9300	C9—H9A	0.9700
C3—C4	1.373 (4)	C9—C8	1.507 (3)
C4—H4	0.9300	N2—N1	1.4191 (17)
C5—H5	0.9300		
C2—C1—H1	120.2	C6—C5—C4	119.07 (19)
C6—C1—H1	120.2	C5—C6—N1	119.50 (15)
C6—C1—C2	119.50 (19)	C1—C6—N1	119.99 (15)
N1—C10—C9	115.39 (14)	C1—C6—C5	120.50 (15)
O1—C10—C9	122.55 (16)	C8—C7—C11	118.64 (15)
O1—C10—N1	122.02 (16)	N2—C7—C11	117.69 (17)
H11B—C11—H11C	109.5	N2—C7—C8	123.63 (14)
H11A—C11—H11C	109.5	H8A—C8—H8B	108.1
C7—C11—H11C	109.5	C9—C8—H8B	109.6
H11A—C11—H11B	109.5	C7—C8—H8B	109.6
C7—C11—H11B	109.5	C9—C8—H8A	109.6
C7—C11—H11A	109.5	C7—C8—H8A	109.6
C1—C2—H2	120.0	C7—C8—C9	110.26 (15)

C3—C2—H2	120.0	H9A—C9—H9B	107.9
C3—C2—C1	120.08 (19)	C8—C9—H9B	109.2
C2—C3—H3	120.0	C10—C9—H9B	109.2
C4—C3—H3	120.0	C8—C9—H9A	109.2
C4—C3—C2	120.02 (17)	C10—C9—H9A	109.2
C5—C4—H4	119.6	C10—C9—C8	112.02 (17)
C3—C4—H4	119.6	N2—N1—C6	113.57 (12)
C3—C4—C5	120.8 (2)	C10—N1—C6	121.30 (13)
C4—C5—H5	120.5	C10—N1—N2	124.06 (13)
C6—C5—H5	120.5	C7—N2—N1	117.09 (14)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C1–C6 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C3—H3 \cdots O1 ⁱ	0.93	2.54	3.371 (3)	149
C5—H5 \cdots O1 ⁱⁱ	0.93	2.50	3.346 (2)	152
C8—H8B \cdots O1 ⁱⁱⁱ	0.97	2.55	3.474 (3)	159
C11—H11B \cdots Cg1 ^{iv}	0.96	2.89	3.812 (3)	161

Symmetry codes: (i) $-x+1, y-1/2, -z+2$; (ii) $x-1, y, z$; (iii) $-x+1, y-1/2, -z+1$; (iv) $-x, y+1/2, -z+1$.

Fig. 1

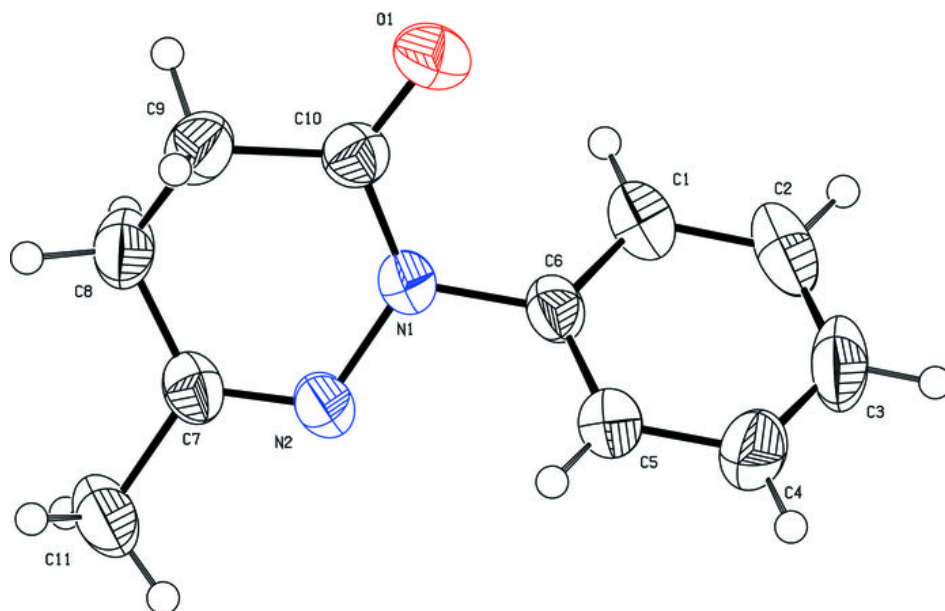


Fig. 2

